Nonradiative quenching of EPR signals in germanium-doped AlGaN: evidence for *DX*-center formation

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(Dated: 21 November 2024)

We present photo-electron paramagnetic resonance (EPR) measurements and first-principles calculations that indicate germanium (Ge) is a DX center in AlGaN. Our photo-EPR measurements on Ge-doped AlGaN samples show no EPR spectra in the dark, while persistent EPR spectra is observed upon photoexcitation with photon energies greater than ~1.3 eV. Thermally annealing the samples decreased the EPR signal, with the critical temperature to quench the EPR signal being larger in the lower Al-content sample. Using detailed first-principles calculations of Ge in AlGaN, we show all of these observations can be explained by accounting for the DX configuration of Ge in AlGaN.

Aluminum nitride (AlN) and high-Al-content $Al_x Ga_{1-x}N$ (AlGaN) alloys are key materials for the future of high-power electronics,¹ high-temperature electronics² and ultra-violet optoelectronics.³ Enabling these technologies is predicated on the ability to achieve controlled doping.⁴ Germanium (Ge) has been explored as a potential *n*-type dopant, motivated in part by experiments that have demonstrated high solubility of Ge in GaN.⁵⁻⁸

One challenge to *n*-type doping of ultra-wide band gap semiconductors such as high-Al-content AlGaN alloys is compensation, either through point defects that act as acceptors or the formation of *DX* centers.^{4,9,10} A *DX* center forms when a large lattice relaxation around a donor dopant leads to the capture of two electrons. This stabilizes the negative charge state with the neutral charge state of the donor being metastable. Hence, *DX* centers inhibit efforts to achieve *n*-type doping, since the dopant effectively self-compensates by establishing the Fermi level in the gap near the resulting (+/-) *DX* level. Prior first-principles calculations predicted a critical Al composition of 53% at which Ge transitions from being a shallow donor to a *DX* center in AlGaN.¹⁰

Several experimental studies have found evidence of carrier compensation in Ge-doped AlGaN, manifesting as a sharp drop in the free carrier concentration.^{11–16} The origin of this observation has been debated: some studies have suggested this occurs due to compensation either by acceptor impurities such as carbon or cation vacancies,¹⁵ while other studies have suggested that Ge leads to a deep donor (+/0) level in the gap.^{14,16} Establishing the properties of Ge in AlGaN, and answering the question of whether it is a *DX* center or not is especially pertinent, given that recent studies have also suggested that AlN could be *n*-type doped with Ge.¹⁷

One possible reason for this ambiguity is that transport measurements alone do not offer a direct way to deduce the stability of the neutral charge state of the Ge donor (Ge_{cation} ; i.e., Ge occupying the cation site in AlGaN or AlN). In this study we present a combination of photoEPR studies and first-principles calculations that establish Ge_{A1} as a DX center in $Al_xGa_{1-x}N$ for x greater than ~0.5. Our photo-EPR measurements identify evidence of the metastable neutral donor following photoexcitation. This EPR signal remains persistent at low temperatures. Performing *in-situ* annealing shows the temperature to quench the EPR spectra decreases as the Al content increases, contrary to what one expects for a deep donor in AlGaN. Using detailed first-principles calculations we show all of these observations are consistent with Ge_{A1} being a DX center.

Two Ge-doped $Al_xGa_{1-x}N$ samples with x=0.65 and x=0.5, and a Si-doped sample with x=0.65, were used in this study. Since our measurements are sensitive to the entire sample, we verified that the signal of interest originated in the AlGaN film by studying a 3.4 μ m thick AlN film deposited on a sapphire substrate. No response was observed under any of the experimental conditions that we use in this study. The AlGaN samples were grown by MOCVD on *c*-oriented sapphire.¹⁸ Each of the samples had a 500 nm layer of undoped AlN deposited directly on sapphire onto which 500 nm of Ge-doped or Si-doped AlGaN was grown. The dopant concentration, measured by secondary ion mass spectrometry (SIMS) is listed in Table I, along with carrier concentrations obtained from Hall measurements.

TABLE I. Aluminum content, x, in Al_xGa_{1-x}N and the dopant in square brackets, dopant concentration measured by secondary ion mass spectroscopy (SIMS), and electron concentration measured at room temperature using Hall measurements reported in Ref 18.

x [Dopant]	Dopants (cm^{-3})	Electrons (cm^{-3})
65% [Si]	1.0×10^{19}	1.0×10^{19}
65% [Ge]	1.0×10^{19}	2.7×10^{16}
$50\%~[{\rm Ge}]$	$3.0{ imes}10^{19}$	1.0×10^{18}

The EPR experiments were done with an X-band spectrometer at 9.4 GHz, which was set up for low temperature (4-300 K) experiments using a liquid helium gas flow apparatus, cryostat, and temperature controller. The samples were aligned with the c axis perpendicular to the magnetic field, to avoid any signal due to the sapphire substrate. The sample was illuminated with LEDs through slits in the microwave cavity. These slits can be covered to ensure the sample was in darkness when necessary, and the photon flux of the LEDs was controlled using neutral density filters. The concentration of neutral donors was determined by calculating the double integral of the EPR spectrum and then comparing this value to one obtained from a heavily phosphorus-doped Si powder reference. To identify the neutral donor state, we determined the g-value which was calculated using $g = C \frac{\nu}{B_0}$, where ν is the microwave frequency, B₀ is the magnetic field where the EPR signal passes through zero, and Cis a correction factor found using the phosphorus-doped Si powder which has a known g-value. Ideally, $C = \frac{h}{\mu_B}$, where h is Planck's constant (4.4×10⁻⁶ eV/GHz) and μ_B is the Bohr magneton (5.79×10⁻⁹ eV/GHz). Measurement of the Si standard typically yields a value for C of 715 Gauss/GHz.

In-situ annealing experiments were done for the Gedoped samples. The sample was illuminated with an LED at 25 K until the EPR signal saturated. The LED was then turned off and an initial EPR scan was taken. The temperature was raised to a target temperature while the sample remained in the cavity; the temperature was held constant for five minutes then lowered back to 25 K, where a subsequent EPR scan was taken. The raising and lowering of the temperature was repeated until the EPR signal was thermally quenched.

To interpret these results we perform first-principles calculations based on density functional theory (DFT),^{19,20} using the projector-augmented wave (PAW) potentials²¹ as implemented in the Vienna Ab-initio Simulation Package (VASP) code.^{22,23} We use the Heyd-Scuseria-Ernzerhof (HSE) hybrid functional²⁴ for all of our calculations. The energy cutoff for the plane-wave basis set is 500 eV. The fraction of nonlocal Hartree-Fock exchange is set to 0.33 for AlN; this results in band gaps and lattice parameters that agree with the experimental values. Defect formation energies and thermodynamic transition levels are calculated using the standard supercell approach 25 with 96-atom supercells. The lattice parameters of the supercell are held fixed while the atomic coordinates are relaxed until the forces are below 5 meV/Angstrom.

To describe the electronic structure of the alloy, the lattice parameters are scaled in accordance with Vegard's law to match the lattice parameters for a given Al content in AlGaN. This approach has previously been successfully applied to describe defects in indium containing GaN alloys.^{26,27} The results are comparable to supercell calculations where Al and Ga atoms are explicitly included into a 96-atom supercell to match a given alloy

composition, and to virtual crystal approximation (VCA) calculations of Ge in AlGaN (see Supplementary Material) The formation energies and charge-state transition levels of Ge are determined using a single $(\frac{1}{4}, \frac{1}{4}, \frac{1}{4})$ k-point. Spin polarization is included for all of our calculations. To analyze the nonradiative and optical properties we construct one-dimensional configuration coordinate diagrams using the Nonrad code.²⁸

Figure 1 illustrates the EPR signal measured at 25 K for the three samples prior to and after illumination. For

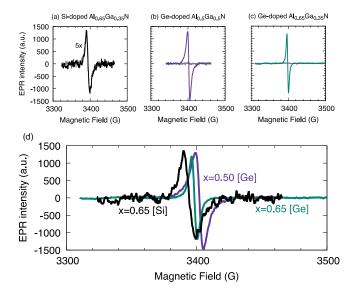


FIG. 1. EPR spectra measured in the dark (grey) and postillumination in Si and Ge-doped $Al_xGa_{1-x}N$. (a) Si-doped $Al_xGa_{1-x}N$ for x=0.65 (multiplied by a factor of 5 for clarity), (b) Ge-doped $Al_xGa_{1-x}N$ for x=0.5 and (c) Ge-doped $Al_xGa_{1-x}N$ for x=0.65. All of the measurements are taken at 25 K. (d) EPR spectra from (a-c) obtained post illumination.

the Si-doped sample we observed an EPR signal in the dark, and the spectrum was unaffected by subsequent illumination, as expected with Si being a shallow donor in AlGaN with 65% Al. For the Ge-doped samples we did not observe any EPR signal above our detection limit of 10^{15} cm⁻³ in the dark. The threshold to photoexcite the EPR signal was 1.3 eV in the Ge-doped samples. The spectra shown in Fig. 1 were obtained using larger photon energies so that a clearer signal could be presented. The EPR signal in the Ge-doped samples also persisted for at least an hour post-illumination.

The photo-induced EPR spectra are replotted in Fig. 1(d) to emphasize the different g-values. The EPR signal in all three samples corresponds to an isotropic g-value of ~1.98. Notably, the g-value for the spectra in the Si-doped sample with 65% Al differs from that in the Ge-doped sample of similar Al content. This difference is possibly due to the different local strain induced by each dopant.¹² The magnitude of the measured g-values and the trend of increasing g-values with increasing Al content (Table II and Fig. 1(d)) are in agreement with prior studies on AlGaN alloys.²⁹ The Ge-doped Al_{0.5}Ga_{0.5}N

TABLE II. Experimentally determined *g*-values from our measurements on Si and Ge-doped AlGaN compared against results in Ref 29.

Al	Dopant	g	Reference
65%	\mathbf{Si}	1.9791(1)	Current work
50%	Ge	1.9749(3)	Current work
65%	Ge	1.9777(1)	Current work
52%	Si	1.9733	Ref. 29
75%	Si	1.9802	Ref. 29

sample had the largest zero-crossing which results in the lowest g-value of all three samples. The linewidths of the three samples ranged from 5-8 G at 25 K. Increasing temperature led to linewidth narrowing. The signals also did not saturate with microwave power up to 10 mW. Each of these characteristics are consistent with the EPR spectra arising from neutral donors.³⁰

For the two Ge-doped samples where the EPR spectra persisted in the dark we conducted *in-situ* annealing experiments and monitored the change in the EPR spectra. In Fig. 2, the change in spin density after annealing for five minutes at a given temperature is shown relative to the initial spin density prior to annealing experiments. The initial spin density in the 50% Al sample is 2.3×10^{17}

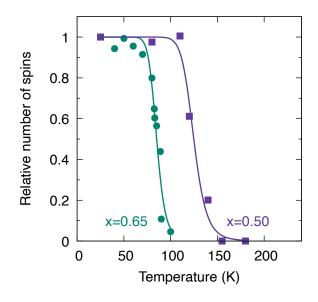


FIG. 2. Relative number of spins obtained from EPR measurements of Ge-doped $Al_x Ga_{1-x}N$ for x=0.65 (green circles) and x=0.50 (purple squares) measured at 25 K after being annealed at specified temperature for five minutes. The size of the symbols is larger than any uncertainty in the temperature. The solid lines are a fit to extract the effective capture barriers. See main text for further details on the fitting.

 $\rm cm^{-3}$ and it is $4.8 \times 10^{16} \rm \ cm^{-3}$ in the 65% Al sample. The number of spins in both samples remains fairly constant

for low anneal temperatures beyond which the number rapidly decreases to a point where the EPR signal disappears and no spins can be detected. For the 50% AlGaN sample, the EPR signal is completely quenched at 160 K, while it is quenched at 100 K for the sample with 65% Al.

Our experiments on the Ge-doped AlGaN samples lead to three principal observations that require a microscopic explanation: the lack of an EPR signal in the Ge-doped samples when measured in the dark, the observation of a neutral donor EPR signal only upon photoexcitation with energies greater than 1.3 eV, and finally the quenching temperature of the EPR signal decreasing with increasing Al content. Using first-principles calculations we highlight how each of these observations are traits of DX centers.⁹

Consistent with previous first-principles studies¹⁰ we find Ge substituting on the Al site is a DX center. The (+/-) level of Ge_{Al} is 1.06 eV below the AlN CBM, while the metastable (0/-) level is 1.57 eV below the AlN CBM, as illustrated within the inset of Fig. 3. In

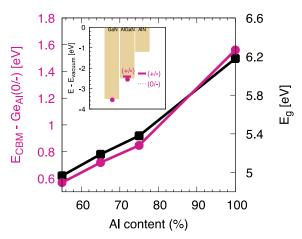


FIG. 3. Energy difference between the $\text{Ge}_{\text{cation}}(0/-)$ level and the AlGaN CBM (magenta-circles, left vertical axis) and the as calculated AlGaN band gap ((black-squares, right vertical axis) plotted as a function of Al content. The inset illustrates the (+/-) and (0/-) level of Ge in AlN and Al-GaN with respect to the conduction band minimum. In the 50% AlGaN alloy and in GaN, Ge is a shallow donor which we denote using the magenta colored circle.

AlN, the metastable neutral charge state, Ge_{Al}^{0} , leads to a breathing-like relaxation of the N ions that are nearestneighbor to Ge_{Al} , resulting in Ge–N bonds that are equivalent in length. In the DX state, Ge_{Al}^{-} , there is a large relaxation of the axial N away from Ge, while the three basal plane Ge-N bond lengths are equivalent in length. We calculated the formation energies and thermodynamic transition levels for Ge substituting on the cation site in AlGaN for x=0.50, 0.55, 0.65 and 0.75 using strained supercell calculations.

For x=0.50, Ge_{cation} acts as a shallow donor and the (+/-) level is above the AlGaN CBM. For x=0.55, 0.65 and 0.75, Ge is stable as a DX center with (+/-) levels

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that are below the AlGaN CBM. The energy difference between the (+/-) level and the AlGaN CBM is 0.07 eV (x=0.55), 0.22 eV (x=0.65) and 0.35 eV (x=0.75) for these intermediate Al compositions. The energy difference between the metastable (0/-) level and the AlGaN CBM also increases with Al content (Fig. 3). The atomic relaxations for each of these alloy compositions are similar to the case of Ge_{Al} in AlN; we find a breathing-like relaxation for Ge⁰_{cation} and Ge⁺_{cation}, while for Ge⁻_{cation} we find a large outward displacement of the axial N.

The EPR-active Ge^0_{cation} state is metastable for x=0.55, 0.65 and 0.75, and 1 in our calculations, explaining why we do not observe any EPR spectra in the dark (Fig. 1(b-c)) in the Ge-doped samples. Photoionization of the DX center above a critical photon energy would convert the EPR-silent Ge_{cation}^- to Ge_{cation}^0 state and activate the EPR signal. We calculate configurationcoordinate diagrams to describe this optical absorption process for x=0.55, 0.65, 0.75 and 1.0 (Fig. 4(a-d)). The peak optical absorption energy based on the Franck-Condon approximation ranges from ~ 1.5 eV (for x=0.55, 0.65, 0.75) to ~ 2.5 eV (x=1). The large lattice relaxation between the negative to the neutral charge state leads to peak optical absorption energies that are significantly larger than dE (Fig. 3). The large peak optical absorption energy (relative to dE) is consistent with our experiments where we find photon energies greater than ~ 1.3 eV are required to photoionize the negative charge state of Ge_{cation} and activate the EPR signal.

Our calculated configuration-coordinate diagrams makes clear that for the x=0.55, 0.65 and 0.75 alloys there is a finite barrier, ΔE_b , to convert from the EPR active $\text{Ge}_{\text{cation}}^0$ state back to the EPR silent $\text{Ge}_{\text{cation}}^-$ state. Because we perform our EPR measurements at low temperature (25 K), there is insufficient thermal energy postillumination for a nonradiative capture process to occur, explaining why the EPR signal remains persistent.

It might be tempting to interpret the subsequent thermal quenching of the EPR signal as the ionization of the Ge_{cation} level. Within this interpretation, the difference in temperature required to quench the EPR signal as the Al content changes may appear to correspond to changes in the depth of the Ge level with respect to the CBM. However, dE increases with Al content (Fig. 3), which is at odds with the higher Al content sample requiring a lower anneal temperature. Rather, we propose this trend in quenching temperatures as a function of Al content corresponds to changes in the nonradiative capture barrier that converts the neutral to the negative charge state. In Fig. 4(e), we illustrate ΔE_b as a function of Al content obtained from our first-principles calculations of the configuration-coordinate diagrams.

To compare the calculated ΔE_b with our EPR quenching measurements, we interpret the thermal quenching of the relative number of spins as nonradiative capture of the photo-excited electrons in the CBM into the Ge_{cation} (0/-) level. We fit our temperature-dependent EPR data (Fig. 2) to a Mott-Seitz type expression for nonradiative capture,³¹ $\frac{n(T)}{n} = \frac{1}{1+\alpha e^{[-E_0/k_BT]}}$, where *n* is the number of spins at the lowest temperature (25 K) that we perform our measurements, α is a frequency factor, and E_0 is the experimental effective capture barrier for the nonradiative process. Keeping α fixed at 10⁸ we find $E_0 =$ 0.198 eV for x=0.5 and $E_0 = 0.135$ eV for x=0.65, which is illustrated in Fig. 2 and Fig. 4(e). If we allow α and E_0 to vary in our fitting, we find $\alpha=2\times10^7$ and $E_0 =$ 0.182 eV for x=0.5, and $\alpha=5\times10^8$ and $E_0 = 0.147$ eV for x=0.65. Consistent with our calculations, we find the capture barrier extracted from our fitting decreases as the Al content increases.

It might be surprising to note that ΔE_b decreases as the Al content increases, since one may assume that ΔE_b increases with dE. However, If $E_{\rm rel}$ is greater than dE, ΔE_b decreases as dE increases. This is apparent if we consider an analytical expression for ΔE_b $\frac{(dE-E_{\rm rel})^2}{4E}$, assuming two identical harmonic oneas $4E_{\rm rel}$ dimensional configuration-coordinate diagrams.³² Our calculations of the configuration-coordinate diagrams as a function of Al content (Figs. 4(a-d)) clarify that E_{rel} for the negative charge state is always large compared to dE. The large relaxation energy is due in part to the lattice relaxation associated with the DX state, explaining why the quenching temperature of the EPR signal decreases for larger Al content.

The experimental and theoretical results presented in this study provide clear evidence that Ge in AlGaN acts as a DX center for Al contents greater than $\sim 50\%$. Our photo-EPR experiments exhibit two traits of Ge_{Al} being a DX center: the absence of an EPR signal in the dark, consistent with the neutral charge state of Ge being metastable, and also photoexcitation activating the neutral donor EPR signal that remains persistent at low temperature. Heating the sample quenches the persistent EPR signal. Surprisingly, we find the temperature to quench the EPR signal decreases with increasing Al content. Using detailed first-principles calculations we show this difference in temperature corresponds to a reduction in the magnitude of the capture barrier associated with the nonradiative process in the EPR thermal quenching measurement. The increase in anneal temperature with deceasing Al content seen in Fig. 2 is a consequence of the large lattice relaxation associated with Ge_{cation} being a DX center.

See the supplementary material for additional details on the alloy calculations.

ACKNOWLEDGMENTS

The work at UAB was supported as part of the Ultra Materials for a Resilient Energy Grid, an Energy Frontier Research Center funded by the U.S. Department of Energy, Office of Science, Basic Energy Sciences under Award No. DE-SC0021230 D.W and J.L.L were supported by the Office of Naval Research through the Naval

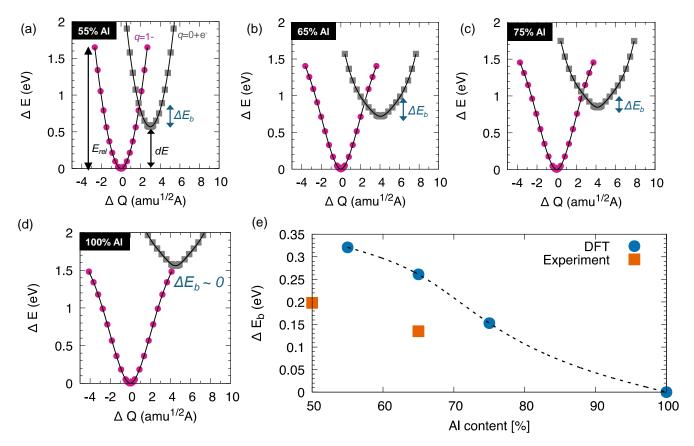


FIG. 4. Configuration coordinate diagrams for $\text{Ge}_{\text{cation}}$ in AlGaN alloys with (a) 55% Al, (b) 65% Al, (c) 75% Al, and (d) 100% Al. The relaxation energy of the initial state, E_{rel} and the ionization energy with respect to the AlGaN CBM, dE is illustrated in (a). The values for dE are from Fig. 3. The classical capture barrier, ΔE_b for each composition is illustrated with a blue vertical arrow. (e) Change in ΔE_b as a function of Al content obtained from panels (a-d) plotted with blue circles. The effective capture barrier by fitting Fig 2 is illustrated with orange squares. See main text for details on the fitting.

Research Laboratory's Basic Research Program. The calculations were supported in part by the DoD Major Shared Resource Center at AFRL.

DATA AVAILABILITY

The data that supports the findings of this study are available from the corresponding author upon reasonable request.

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